Figure: 30 TAC 106.492(1)(D)

For HCl $Q = 2.73 \times 10^5 \times HCl$

For SO_2 $Q = 0.53 \times 10^5 \times SO_2$

Where Q = heat release, British thermal units per hour, based on lower heating value

HCl = HCl emission rate, lb/hr

SO_2 = SO_2 emission rate, lb/hr